

## 2-Amino-4,6-dimethylpyridinium benzoate

Mohd Razip Asaruddin,<sup>a</sup> Habibah A Wahab,<sup>b‡</sup> Nornisah Mohamed,<sup>a</sup> Mohd Mustaqim Rosli<sup>c</sup> and Hoong-Kun Fun<sup>c\*§</sup>

<sup>a</sup>Pharmaceutical Design and Simulation Laboratory, School of Pharmaceutical Sciences, Universiti Sains Malaysia, 11800 USM, Penang, Malaysia, <sup>b</sup>Institute of Pharmaceutical and Neutraceuticals, Malaysia Ministry of Science and Technology and Innovation, Science Complex, 11900, Malaysia, and <sup>c</sup>X-ray Crystallography Unit, School of Physics, Universiti Sains Malaysia, 11800 USM, Penang, Malaysia  
Correspondence e-mail: hkfun@usm.my

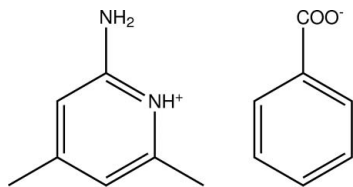
Received 17 August 2010; accepted 29 August 2010

Key indicators: single-crystal X-ray study;  $T = 296$  K; mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å;  $R$  factor = 0.044;  $wR$  factor = 0.127; data-to-parameter ratio = 11.7.

In the title compound,  $\text{C}_7\text{H}_{11}\text{N}_2^+ \cdot \text{C}_7\text{H}_5\text{O}_2^-$ , the 2-amino-4,6-dimethylpyridinium cation and the benzoate anion are linked by two  $\text{N}-\text{H} \cdots \text{O}$  hydrogen bonds, forming an  $R_2^2(8)$  ring motif. The H atoms in both the methyl groups are rotationally disordered, with fixed site occupancies of 0.50. In the crystal structure, the molecules are stabilized by intermolecular  $\text{N}-\text{H} \cdots \text{O}$  hydrogen bonds. A  $\pi-\pi$  interaction, with a centroid-centroid distance of 3.661 (2) Å, is also observed.

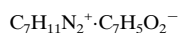
### Related literature

For the biological activity of Schiff bases with azomethine linkages, see Dhar & Taploo (1982). For hydrogen bonding, see: Jeffrey (1997); Jeffrey & Saenger (1991). For graph-set descriptions of hydrogen-bond ring motifs, see: Bernstein *et al.* (1995).



### Experimental

#### Crystal data



$M_r = 244.29$

<sup>‡</sup> Additional correspondence author, e-mail: habibah@ipharm.gov.my or habibahw@usm.my. Institute of Pharmaceutical and Neutraceuticals, Malaysia Ministry of Science and Technology and Innovation Science Complex, 11900, Penang, Malaysia.

<sup>§</sup> Thomson Reuters ResearcherID: A-3561-2009.

Monoclinic,  $P2_1/c$   
 $a = 7.5362$  (16) Å  
 $b = 22.937$  (4) Å  
 $c = 8.2124$  (14) Å  
 $\beta = 109.820$  (2)°  
 $V = 1335.5$  (4) Å<sup>3</sup>

$Z = 4$   
Mo  $K\alpha$  radiation  
 $\mu = 0.08$  mm<sup>-1</sup>  
 $T = 296$  K  
 $0.57 \times 0.23 \times 0.05$  mm

#### Data collection

Bruker SMART APEXII CCD  
area-detector diffractometer  
Absorption correction: multi-scan  
(*SADABS*; Bruker, 2009)  
 $T_{\min} = 0.954$ ,  $T_{\max} = 0.996$

7639 measured reflections  
2336 independent reflections  
1527 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.029$

#### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.044$   
 $wR(F^2) = 0.127$   
 $S = 1.02$   
2336 reflections  
199 parameters

H atoms treated by a mixture of independent and constrained refinement  
 $\Delta\rho_{\text{max}} = 0.10$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.14$  e Å<sup>-3</sup>

Table 1

Hydrogen-bond geometry (Å, °).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$\text{N1}-\text{H1N1} \cdots \text{O2}$	1.04	1.65	2.683 (2)	172
$\text{N2}-\text{H1N2} \cdots \text{O1}$	1.01	1.78	2.779 (2)	171
$\text{N2}-\text{H2N2} \cdots \text{O1}^i$	0.90	1.97	2.853 (2)	168

Symmetry code: (i)  $x, -y + \frac{1}{2}, z - \frac{1}{2}$ .

Data collection: *APEX2* (Bruker, 2009); cell refinement: *S SAINT* (Bruker, 2009); data reduction: *S SAINT*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 2008); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL* and *PLATON* (Spek, 2009).

This research was supported by Universiti Sains Malaysia (USM) under the University Research grant (No. 1001/PFARMASI/815004) and the Ministry of Science, Technology and Innovation through an R&D Initiative Grant (09-05-IFN-MEB 004). HKF and MMR also thank USM for the Research University Grant (No. 1001/PFIZIK/811160). MRA gratefully acknowledges a PhD scholarship from Universiti Malaysia Sarawak.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: FJ2327).

### References

- Bernstein, J., Davis, R. E., Shimon, L. & Chang, N.-L. (1995). *Angew. Chem. Int. Ed. Engl.* **34**, 1555–1573.  
Bruker (2009). *APEX2*, *S SAINT* and *SADABS*. Bruker AXS Inc., Madison, Wisconsin, USA.  
Dhar, D. N. & Taploo, C. L. (1982). *J. Sci. Ind. Res.* **41**, 501–506.  
Jeffrey, G. A. (1997). *An Introduction to Hydrogen Bonding*. Oxford University Press.  
Jeffrey, G. A. & Saenger, W. (1991). *Hydrogen Bonding in Biological Structures*. Berlin: Springer.  
Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.  
Spek, A. L. (2009). *Acta Cryst.* **D65**, 148–155.