



Performance of TiO₂/Ag/CA for efficient adsorption of methylene blue

Devagi Kanakaraju¹ · Feniellia Diwvya Anak Kutiang¹ · Sheikh Mohammad Fareth Abdul Hadi¹ · Ying Chin Lim²

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Abstract

The incorporation of additives and inclusion of metal dopants on titanium dioxide (TiO₂) could improve its inherent limitations and result in enhanced photocatalytic efficiency. In this work, the performance of TiO₂/Ag with the use of cellulose acetate (CA) was investigated for the degradation of a model dye pollutant under different operational settings. The surface morphology, elemental composition, crystallinity, and band gap of TiO₂/Ag/CA composite were verified by scanning electron microscopy, energy-dispersive X-ray, X-ray diffraction, and UV–Vis diffuse reflectance spectroscopy, respectively. An increase in CA concentration from 1 to 2 wt% led to the high removal of methylene blue (MB). The highest removal of MB (90.90%) was obtained by TiO₂/Ag/CA under the conditions of 4 g/L catalyst loading, 5 ppm MB concentration, and at unadjusted pH (7.8). The recyclability study showed that the removal efficiency of the retrieved photocatalyst remains as much as ~78% after 3 cycles, demonstrating its sustainability in water treatment. The remarkable adsorption effect shown by CA as an additive greatly enhanced the removal of MB by the TiO₂/Ag photocatalyst under UV irradiation and dark condition. The TiO₂/Ag/CA composite, which has dual roles in adsorption and photocatalysis, could be employed as a pollution removal agent in water.

Keywords Silver · Photocatalysis · Titanium dioxide · Adsorption · Wastewater · Dye