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DYNAMIC MODELLING OF HYDROGEN PRODUCTION FROM PHOTO-FERMENTATION IN MICROBIAL ELECTROLYSIS CELL USING SAGO WASTE

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INTRODUCTION

Diminishing of the conventional fossil fuel reserves in the future

- Hydrogen promising alternative fuel (high efficient energy carrier), viable, sustainable and zero emission combustion [only generate byproduct of water vapour + thermal]
- Currently, reforming of fossil fuel is widely used for the substantial hydrogen supply but technologically not efficient

•Biomass as a promising replacement to fossil fuel for a sustainable bioohydrogen production

RESEARCH PROBLEMS

Scope of Studies

 Sago waste increases due to the rising of sago commercialisation in Sarawak, thus require a good waste management strategy

 MECs alone break down carbohydrates slowly and produce low hydrogen purity

- Use sago waste as a raw materials in the Microbial electrolysis cell (MEC) for biohydrogen production
- Integration of MEC with a photo-fermentation – complete and efficient conversion
- Mathematical modelling reaction and kinetic studies

RESEARCH OBJECTIVES

- To establish mathematical model for bio-hydrogen production in the MEC with the integration of photo-fermentation by using sago's waste
- To study in depth the chemical reaction of bio-hydrogen production in the integrated MEC-photo-fermentation
- To observe the influences of the concentration of microbial community on the hydrogen production at batch process in which including the parameters as following:
 - Substrate concentration
 - Fermentative microorganism concentration
 - Electricigenic microorganism concentration
 - Hydrogenotrophic methanogens concentration
 - Microbial Electrolysis Cell (MEC) current

FINDING AND DISCUSSION



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methane Acetate -→



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CONCLUSION AND FUTURE WORKS

• A mathematical model of the MEC with integrated of photo-fermentation for hydrogen production has been developed and modified by using sago effluent as a substrate in a batch process mode.

- Electricigenic microorganism > acetoclastic methanogens = desirable hydrogen production
- Maximum 3.8 L of hydrogen per day can be generated at MEC current of 0.38 A and hydrogenotropic methanogen concentration of 10 mg/L (less methane is emitted)
- Model is able to predict and correlate between hydrogen production with concentration of electricigenic microorganism and MEC current.
- The model will be further optimized with the main objective to maximize the hydrogen production