

REMOVAL COLOR OF DYES FROM TEXTILE INDUSTRIAL WASTEWATER BY SAGO WASTE AS A COAGULANT

RICKY ANAK TUBAM

Bachelor of Engineering with Honours (Chemical Engineering) 2016/2017

UNIVERSITI MALAYSIA SARAWAK

DECLARATION OF ORIGINAL WORK

This declaration is made on the <u>26 day of May 2017</u>.

Student's Declaration:

I, <u>RICKY ANAK TUBAM (43823)</u>, <u>DEPT. OF CHEMICAL ENGINEERING AND ENERGY</u> <u>SUSTAINABILITY, FACULTY OF ENGINEERING hereby declare that the work entitled, REMOVAL</u> <u>COLOR OF DYES FROM TEXTILE INDUSTRIAL WASTEWATER BY SAGO WASTE AS A</u> <u>COAGULANT</u> is my original work. I have not copied from any other students' work or from any other sources except where due reference or acknowledgement is made explicitly in the text, nor has any part been written for me by another person.

26 MAY 2017 Date submitted

RICKY ANAK TUBAM (43823)

Name of the student (Matric No.)

Supervisor's Declaration:

I <u>SITI HAZIRAH ADAM</u> hereby certifies that the work entitled, <u>REMOVAL COLOR OF DYES FROM</u> <u>TEXTILE INDUSTRIAL WASTEWATER BY SAGO WASTE AS A COAGULANT</u> was prepared by the above named student, and was submitted to the "FACULTY" as a * partial KNC 4322 FINAL YEAR PROJECT 1 fulfillment for the conferment of <u>BACHELOR OF ENGINEERING WITH HONOURS</u> (<u>CHEMICAL ENGINEERING</u>), and the aforementioned work, to the best of my knowledge, is the said student's work

Received for examination by: <u>SITI HAZIRAH ADAM</u> (Name of the supervisor)

Date: 26 MAY 2017

I declare this Report is classified as (Please tick $(\sqrt{})$):

CONFIDENTIAL (Contains confidential information under the Official Secret Act 1972)*

RESTRICTED (Contains restricted information as specified by the organisation where research was done)*

 $\overline{\checkmark} \quad \text{OPEN ACCESS}$

Validation of Report

I therefore duly affirmed with free consent and willingness declared that this said Report shall be placed officially in Department of Chemical Engineering and Energy Sustainability with the abide interest and rights as follows:

- This Report is the sole legal property of Department of Chemical Engineering and Energy Sustainability, Universiti Malaysia Sarawak (UNIMAS).
- The Department of Chemical Engineering and Energy Sustainability has the lawful right to make copies for the purpose of academic and research only and not for other purpose.
- The Department of Chemical Engineering and Energy Sustainability has the lawful right to digitise the content to for the Local Content Database.
- The Department of Chemical Engineering and Energy Sustainability has the lawful right to make copies of the Report for academic exchange between Higher Learning Institute.
- No dispute or any claim shall arise from the student itself neither third party on this Report once it becomes sole property of Department of Chemical Engineering and Energy Sustainability, Universiti Malaysia Sarawak (UNIMAS).
- This Report or any material, data and information related to it shall not be distributed, published or disclosed to any party by the student except with Department of Chemical Engineering and Energy Sustainability, Universiti Malaysia Sarawak (UNIMAS) permission.

Student's signature

(26 MAY 2017)

Supervisor's signature:

(26 MAY 2017)

Current Address:

LOT 896, LORONG 4B2 TAMAN SAMARINDAH, 94300 KOTA SAMARAHAN, SARAWAK

Notes: * If the Report is **CONFIDENTIAL** or **RESTRICTED**, please attach together as annexure a letter from the organisation with the period and reasons of confidentiality and restriction.

APPROVAL SHEET

This final year report which entitled **"Removal Color of Dyes from Textile Industrial Wastewater by Sago Waste as a Coagulant"** was prepared by Ricky anak Tubam (43823) is hereby read and approved by:

> 26 MAY 2017 Date

SITI HAZIRAH ADAM (Final Year Project Supervisor)

REMOVAL COLOR OF DYES FROM TEXTILE INDUSTRIAL WASTEWATER BY SAGO WASTE AS A COAGULANT

RICKY ANAK TUBAM

A dissertation submitted in partial fulfillment of the requirement for the degree of Bachelor of Engineering with Honours (Chemical Engineering and Energy Sustainability)

> Faculty of Engineering Universiti Malaysia Sarawak

> > 2017

Dedicated to my beloved parents, who always there for me give me sustainable motivations, support and encouragements

ACKNOWLEDGEMENT

The author wishes to thanks to all individuals, parties and organizations that have contributed and cooperated throughout this project especially to Ms. Siti Hazirah Adam, lecturer of Department of Chemical Engineering and Energy Sustainability because of her guidance, knowledge, support, and experiences as well as for her invaluable supervision towards the completion of this report. Last but not least, special gratitude is dedicated to all family members and fellow friends for help, support and contribution.

ABSTRACT

The aim of this study was to remove the color of dyes from textile industrial wastewater by sago waste acted as a coagulant. The low-cost, easily available and naturally prepared coagulant like sago waste was chosen to remove color of Congo Red. The experiment was carried out by using jar test experiment. Jar test was used to perform coagulation process. Factors affecting the coagulation and flocculation process, such as pH and dosage were assessed. Fourier Transform Infrared (FT-IR) was used in this study. The purpose of using FT-IR was to examine the formation dyes molecule on the surface of the sago waste before the coagulation process. Three types of coagulant were prepared; untreated sago waste, treated sago waste with sodium hydroxide and treated sago waste with sodium hypochlorite. Next, for data analysis, the dyes percentage removal was measured by using UV-visible spectrophotometer. The optimum condition of each parameter was analyzed. The data was tabulated and discussed.

ABSTRAK

Tujuan kajian ini adalah untuk menyinkirkan pewarna daripada air sisa industri tekstil dengan menggunakan sisa sagu sebagai *coagulant agent*. Sisa sagu ialah berkos rendah, mudah didapati dan disediakan secara semula jadi telah menjadi pilihan sebagai *coagulant agent* untuk meyinkirkan warna *Congo Red*. Ekperimen akan dijalankan dengan menggunakan *jar test*. Jar test akan digunakan untuk mengkaji process penggumpalan. Fourier Transfrom Infared (FT-IR) akan digunakan dalam kajian ini. Tujuan penggunaan FT-IR ini adalah untuk menkaji pewarna pembentukan melekul di permukaan sisa sagu sebelum proses penggumpalan. Terdapat tiga jenis koagulan digunakan dalam kajian ini iaitu sisa sagu yang tidak dirawat, sisa sagu yang dirawat menggunakan natrium hidroksida dan sisa sagu yang dirawat dengan natrium hipoklorit. Seterusnya, untuk analisis data, kecekapan pewarna penyingkiran akan diukur dengan menggunakan spektrofotometer UV-visible. Keadaan optimum bagi setiap parameter dianalisis. Data ini akan dijadualkan dan dibincangkan.

TABLE OF CONTENTS

	Pages
DECLARATION OF ORIGINAL WORK	i
APPROVAL SHEET	iii
TITLE PAGE	iv
DEDICATION	v
ACKNOWLEDGEMENT	vi
ABSTRACT	vii
ABSTRAK	viii
TABLE OF CONTENTS	ix
LIST OF TABLES	xi
LIST OF FIGURES	xii
ABBREVIATIONS	xiii
NOMENCLATURE	xiv

CHAPTER 1	INTRODUCTION		
	1.1	Introduction	1
	1.2	Problem Statement	3
	1.3	Objectives	4
	1.4	Scope of study	4
	1.5	Expected Outcome	4
	1.6	Schedule and Gantt chart	5

CHAPTER 2 LITERITURE REVIEW

2.1	History of Textile Wastewater	7
2.2	Dyes Technologies	8
2.3	Coagulation Process	9
2.4	Overview of Sago	11

CHAPTER 3 METHODOLOGY

3.1	Introduction	13
3.2	Sago Waste Preparation	13

3.3	Wastewater Preparation	16
3.4	Congo Red	16
3.5	Experiment Set-Up	16
3.6	Characterization	18
3.7	Data Analysis	19
3.8	Beer's Law	19

CHAPTER 4 RESULT AND DISCUSSION

4.1	Introduction	21
4.2	Effect of pH	21
4.3	Effect of Dosage	31
4.4	Characterization of the Sago Waste using FT-IR	37

CHAPTER 5 CONCLUSION AND RECOMMENDATION

5.1	Conclusion	42
5.2	Recommendation	43

REFERENCES

APPENDIX

47

44

LIST OF TABLES

Table		Page
1.1	Advantages and disadvantages of dye removal methods	2
1.2	Schedule for completing Final Year Project 1	5
1.3	Schedule for completing Final Year Project II	5
2.1	Wastewater Treatment Methods	8
2.2	Type of aluminium and iron coagulant used in wastewater treatment	10
2.3	Composition of sago waste	12
3.1	The apparatus and material used in this experiment	16
4.1	Titration of Congo Red solution with untreated sago	22
4.2	UV-vis spectrometer reading of Congo Red (untreated sago) with	24
	different pH	
4.3	Titration of Congo Red solution with treated sago (NaOH)	25
4.4	UV-vis spectrophotometer reading of Congo Red (treated sago	26
	NaOH) with different pH	
4.5	Titration of Congo Red solution with treated sago (NaClO)	28
4.6	UV-vis spectrophotometer reading of Congo Red (treated sago	28
	NaClO) with different pH	
4.7	UV-vis spectrophotometer reading of Congo Red (untreated sago)	32
	with different sago dosage	
4.8	UV-vis spectrometer reading of Congo Red (treated sago NaOH)	33
	with different sago dosage	
4.9	UV-vis spectrophotometer reading of Congo Red (treated sago	35
	NaClO) with different sago dosage	

LIST OF FIGURES

Figure		Page
1.1	Gantt chart for completing the Final Year Project 1	6
2.1	General process involve in coagulation-flocculation	10
2.2	Image of Sago Waste	11
3.1a	Untreated sago waste	14
3.1b	Treated sago waste with NaOH	14
3.1c	Treated sago waste with NaClO	15
3.2	Jar Tester	17
3.3	FT-IR used in this study	18
3.4	KBr pellets for FT-IR analysis	18
4.1	Five set of Congo Red solutions with different pH	22
4.2	Picture of performing jar test	23
4.3	Color of Congo Red in pH 5 solution after jar test	23
4.4	The percentage removal of untreated sago waste	24
4.5	Color of Congo Red solution with treated sago waste (NaOH) after jar test	26
4.6	The percentage removal of treated sago waste (NaOH)	27
4.7	The percentage removal of treated sago waste (NaClO)	29
4.8	Influence of pH on Congo Red removal by treated and untreated sago waste as a coagulant	29
4.9	Effect of pH on Congo Red removal by natural coagulant (Patel and Vashi, 2010)	30
4.10	The percentage removal against dosage of untreated sago waste	32
4.11	The color of Congo Red after jar test with treated sago	33

waste (NaOH) as a coagulant

4.12	The percentage removal against dosage of	34
	treated sago waste with NaOH	
4.13	The percentage removal against dosage of treated sago waste with NaClO	35
4.14	Effect of coagulant dose on Congo Red removal.	36
4.15	FT-IR used throughout the analysis of characterization of the sago waste	37
4.16	FT-IR spectrum for untreated sago waste.	38
4.17	FT-IR spectrum for treated sago waste with NaOH	39
4.18	FT-IR spectrum for treated sago waste with NaClO	40
4.19	Comparison of FT-IR for treated and untreated sago waste	41

ABBREVIATIONS

- DOE Department of Environment
- COD Chemical Oxygen Demand
- SEM Scanning Electron Microscope
- FTIR Fourier Transform Infrared
- UNIMAS Universiti Malaysia Sarawak
- NaOH Sodium Hydroxide
- NaClO Sodium Hypochlorite
- KBr Potassium Bromide

NOMENCLATURE

ml	Millimeter
Rpm	Revolutions per minute
cm	Centimeter
Ppm	Parts per million

CHAPTER 1

INTRODUCTION

1.1 Introduction

According to the Malaysia Environmental Quality Report (2014) regulated by Department of Environment (DOE) Malaysia, analyzed based on 473 rivers, it is found that 244 (52%), were cleaned, 186 (39%) were polluted and 43 (9%) were in serious polluted condition. The textile industry has drawn the attention of environment that posed a high demand on water supply and produced large amount of wastewater. The most used dyes in textile industries are Remazol and Vinyl Sulphone fibre reactive dyes (Rashidi et al., 2012). Most reactive dyes in wastewater are usually difficult to biodegrade with high pH, high COD and strong color. Thus, many processes are studied for the removal of reactive dyes by conventional treatment technologies including biological and chemical oxidation, chemical coagulation, photocatalysis and adsorption processes.

There are numerous method to treat dye bearing effluents. In spite of the availability of many techniques to remove dye contaminants from wastewaters, such as coagulation, chemical oxidation, membrane separation process, electrochemical and aerobic and anaerobic microbial degradation, each of these methods have inherent limitations. The technologies can be divided into three categories: physical, chemical and biological. All these method have their own advantages and disadvantages. Table 1.1 shows the advantages and disadvantages of different dye removal methods.

Method	Advantages	Disadvantages
Ozonation	Available used in gaseous state	High capital cost and short half-life which is 20 minutes
Adsorption	Excellent in removing of wide variety of dyes	Difficult to regeneration, costly disposal of adsorbent
Membrane filtration	Able to remove all type of dye	Produce high concentration of sludge and high operating cost
Ion exchange	Easy regeneration	Not effective for all type of dyes
Electro- coagulation	Good in removing dyes	Highly cost and less electrode reliability
Biological process	Environmental friendly	Slow process, need nutrients
Chemical coagulation and flocculation	Excellent color removal and economically feasible	Produce sludge

Table 1.1: Advantages and disadvantages of dye removal methods (Akshaya et al.,

2011)

In this study, the coagulation method will be used to remove the dyes molecules from textile wastewater. Coagulation is one of the most popular unit operation in water and wastewaters treatment trains. It is one of the most effective methods for dye removal from industrial wastewater. Coagulation is easily applied, bulky treatment and low capital and operating costs, being usually employed as pre-treatment (Elizalde and Hernandez, 2009). In industrial effluent applications, two most important parameters which influence the coagulation process are pH, alkalinity and temperature. The optimum pH range varies depending on the coagulants used and is typically determined by laboratory testing commonly known as jar test experiments.

The following are examples of waste materials used in dye removal: coconut shell, corncob wastes and avocado kernel seeds. These were used to remove acid, basic, and reactive dyes. Powdered peanut hull was used as a biosorbent for anionic dyes, guava seeds were tested as an adsorbent for the removal of acid dyes and palm fruit bunch wastes were used for basic dyes, alongside others. However, locally available and abundant low-cost coagulant agent are still necessary in order to remove industrial textile wastewater.

Sarawak is the main producer of sago flour in Malaysia. The Sarawak state is recognized not only as one of the world's biggest sago growing area but also the biggest exporter of sago products mainly in the form of starch, sago flour and sago pith - totalling about 47,000 metric ton (~ 80 RM Million) are export annually to Peninsular Malaysia, Japan, Singapore, Thailand, Taiwan, United States, and other countries (Zayas et al., 2007). Now, the agricultural waste, such as that produced from the sago processing plant contains cellulose that may be exploited to be a new value added product. Work on the potential of sago waste to be converted as a coagulant agent is still considered limited including the study on its physicochemical properties, its structure and its behavior/capability in absorbing the different kind of Textile dyes.

This project is aimed to treat wastewater from textile industries using sago waste coagulant by jar test experiment. Sago waste were dried and blended into powder with particular size between 60-100 micrometer. Dried sago waste was dissolved in distilled water, and then mixed with wastewater to allow the coagulation process. Factors affecting the coagulation and flocculation process, such as pH and dosage will be assessed. The performance of coagulation will be assessed based on the quality of treated wastewater, red in term of reduction dyes removal and Turbidity.

1.2 Problem Statement

Malaysia is one of the countries in the world that consist of highest local sources of sago especially in Sarawak. Sarawak exporting tons of sago products annually to Peninsular Malaysia, Taiwan, Singapore, Japan and other countries. This abundant sago waste can be converted becomes a new value-added product which may become another option to vary the use of sago. The outcome of this study may contribute to the development of sago industries in Sarawak.

Dyes is one of the chemical that was used in textile industry and considered as important pollutant. In Malaysia, batik industry is one of the best textile industry. Unfortunately, the dye wastewater produced from the industry is usually being direct discharge into waterways without any treatment (Khai Ern Lee et al., 2015). Water is one of the raw materials needed in many industries such as textile industry. Textile industry is one of the industries that consumed a huge amount of water. Therefore, the probability of these industry to produce water pollution is higher. The escalating demand of textile products cause the wastewater produce from the textile industry increasing. Textile industry can be known as industry that is focused on the design and production of yarn and cloth. These industry commonly produced wastewater that contain both chemical and dyes.

The chemical used in textile industry cause damage in environment and ecosystem. Therefore, the removal color of dyes from textile industrial wastewater is important to reduce and avoid pollution which can damage the environment. Hence, the treatment color of dyes from textile industrial wastewater by sago waste (solid) act as coagulant is studied in this research.

1.3 **Objectives**

The aim of this study is to remove the color of dyes from textile industrial wastewater by sago waste as a coagulant. From this aim, there are three objectives that will be focused to achieve the aim of the study:

- i. To evaluate the performance of the treated and untreated sago wastes as a coagulant to remove the color of dyes from textile industry.
- ii. To characterize physical-chemical properties of the sago waste as a coagulant agent
- iii. To analyze the optimum condition and the capability of sago waste to treat color of dyes by coagulation method.

1.4 Scope of Study

The following scopes for this research were identified in order to achieve the objectives.

- i. Characterization of the Sago Waste Coagulant by Fourier Transform Infrared (FTIR).
- ii. Evaluate the performance of the sago waste (coagulant) on industrial textile dye wastewater under various parameters such as pH and coagulant agent dosage.

1.5 Expected outcome

 Abundant Sago waste will be converted become a new value-added product, coagulant agent, which may become another option to vary the use of sago. The outcome of this study may contribute to the development of sago industries in Sarawak.

- ii. Sago wastes coagulant can be adopted as a low-cost coagulant agent by coagulation method in removal of industrial textile wastewater.
- iii. Replace the existing chemical used in industrial textile wastewater by introducing low-cost abundant Sago waste coagulant.
- iv. Propose a cost effective method of treating industrial wastewater containing dyes color, and hope it will help in industry textile Malaysia to meet the standard of sewage discharged set by Department of Environment (DOE), Malaysia.

1.6 Schedule and Gantt chart

TASK	START	FINISH	DURATION				
Introduction to FYP 1	05/09/2016	09/09/2016	5 days				
Chapter 1	12/09/2016	23/09/2016	10 days				
Chapter 2	26/09/2016	21/10/2016	20 days				
Chapter 3	24/10/2016	18/11/2016	20 days				
Preparation for presentation FYP 1	21/11/2016	30/11/2016	8 days				

Table 1.2: Schedule for completing Final Year Project I

Table 1.3 : Schedule for completing Final Year Project II

TASK	START	FINISH	DURATION					
Experiment s and Analysis	09/03/2017	09/05/2017	2 months					
Chapter 4 writing	09/05/2017	24/09/2017	15 days					
Chapter 5 writing	24/05/2017	25/05/2017	1 day					
Preparation for presentation FYP II	26/05/2017	29/05/2017	3 days					

p Task Name	Sept 2016					Okt 2016									Nov 2016									
INTRODUCTION TO FINAL YEAR PROJECT 1																								
2 CHAPTER 1			_																	v				
3 CHAPTER 2					i h		_																	
4 CHAPTER 3				8										_		_								
5 PRSENTATION FINAL YEAR 5 PROJECT 1																								

Figure 1.1 : Gantt chart for completing the Final Year Project 1

CHAPTER 2

LITERITURE REVIEW

2.1 History of Textile Wastewater

In Malaysia, Batik industry is one of the most common textile industry. Textile industry basically used cotton, wool, silk, and polymer fibers as their main raw materials. In order to fulfill human standard of living and fashion, textile industry tends to produce and release more hazardous compounds. Textile industry is one of the most environmentally unfriendly industrial due to the colored wastewater produced that are heavily polluted with dyes and chemicals. According to Azam and Mohammad (2013), it is approximated that 5000 tons of dyeing material from textile industry are released into environment every year and the contaminated materials are absorbed by the oxygen of the water. The wastewater released from the textile industry can pollute the underground water. The wastewater was contained of toxic substances such as cyanides, alkaline cleaning agent, degreasing solvent, fat, oil and metals which can threatens human life and the environment.

Wastewater treatment methods can be classified into physical, chemical and biological methods according to the nature of the treatment process that is used as shown in Table 2.1. Physical methods are generally processes that do not show any chemical or biological changes and used to enhance the wastewater. Chemical method generally involve in some chemical reactions to improve the quality of wastewater. Biological method use microorganisms to treat the wastewater but usually biological method not able to degrade complicated dyes and take a long period of time.