

WATER AND SEDIMENT QUALITY OF MURUM RESERVOIR, INUNDATED PLIERAN RIVER AND DANUM RIVER

Toh Yu Mei

Bachelor of Science with Honours (Resource Chemistry) 2015

Water and Sediment Quality of Murum Reservoir, Inundated Plieran River and Danum River

Toh Yu Mei

39183

A report submitted in partial fulfilment of the requirements for the degree of Bachelor of Science (Hons.)

Supervisor: Assoc. Prof. Dr. Ling Teck Yee Co-supervisor: Prof. Lee Nyanti

> Resource Chemistry Chemistry Department

Faculty of Resource Science and Technology Universiti Malaysia Sarawak 2015

Acknowledgements

I would like to say thank you to my supervisor, Assoc. Prof. Dr. Ling Teck Yee, for her advice and guidance throughout this research. Besides, I would like to thank Prof. Dr. Lee Nyanti for his help and guidance during field trips. I appreciate the financial support from the Sarawak Energy Berhad through Grant No. GL (F07)/SEB/4C/2013(26).

Moreover, I also would like to thank to Universiti Malaysia Sarawak for providing the equipment and facilities. I would like to thank the laboratory assistants for their guidance in using the chemicals and instruments. In addition, I would like to thank my fellow friends who help throughout this research. Last but not least, I would like to thank my parents for their support mentally.

Declaration

I hereby declare that this thesis entitled 'Water and Sediment Quality of Murum Reservoir, Inundated Plieran River and Danum River' is an original research work of my own excluded for the references document. The report has not been accepted for any degree and is not concurrently submitted for any other degree.

(TOH YU MEI)

Date:

Resource Chemistry Programme

Department of Chemistry

Faculty of Resource Science and Technology

Universiti Malaysia Sarawak

Acknowledgements	II
Declaration	III
Table of Contents	IV
List of Abbreviations	VIII
List of Tables	X
List of Figures	XII
List of Appendices	XV
ABSTRACT	1
1.0 Introduction	2
2.0 Literature review	4
2.1 Water Quality Parameters	4
2.1.1 Temperature	4
2.1.2 pH	4
2.1.3 Dissolved Oxygen (DO)	5
2.1.4 Electrical conductivity (EC)	6
2.1.5 Turbidity	6
2.1.6 Transparency	7
2.1.7 Five-day Biochemical Oxygen Demand (BOD5)	7
2.1.8 Total Suspended Solids (TSS)	8
2.1.9 Nitrite-Nitrogen (NO ₂ -N) and Nitrate-Nitrogen (NO ₃ -N)	8
2.1.10 Total Ammonia Nitrogen (TAN)	9
2.1.11 Total Nitrogen (TN)	9
2.1.12 Soluble Reactive Phosphorus (PO ₄ -P)	9
2.1.13 Total Phosphorus (TP)	10
2.1.14 Total Sulfide (TS ²⁻)	10
2.1.15 Chemical Oxygen Demand (COD)	11
2.1.16 Chlorophyll-a (chl-a)	11
2.2 Sediment Quality	12
2.3 Sediment Sample Parameters	12
2.3.1 pH	12

Table of Contents

2.3.2 Water Content	
2.3.3 Organic Matter	
2.3.4 Total Organic Carbon (TOC)	
2.3.5 Particle Size Analysis (PSA)	
2.3.6 Total Nitrogen (TN)	14
2.3.7 Total Phosphorus (TP)	14
3.0 Materials and Methods	
3.1 Study Area	
3.2 Sample Preparation and Collection	16
3.3 Water Quality In-situ Measurements	
3.4 Water Quality <i>Ex-situ</i> Measurements	
3.4.1 Five-day Biochemical oxygen demand (BOD ₅)	
3.4.2 Total suspended solids (TSS)	
3.4.3 Nitrite-Nitrogen (NO ₂ -N)	
3.4.4 Nitrate-Nitrogen (NO ₃ -N)	
3.4.5 Total Ammonia Nitrogen (TAN)	
3.4.6 Total Nitrogen (TN)	
3.4.7 Soluble Reactive Phosphorus (PO ₄ -P)	22
3.4.8 Total Phosphorus (TP)	
3.4.9 Total Sulfide (TS ²⁻)	23
3.4.10 Chemical Oxygen Demand (COD)	24
3.4.11 Chlorophyll-a (Chl-a)	25
3.5 Sediment Quality Measurements	
3.5.1 pH	
3.5.2 Water Content (W)	
3.5.3 Organic Matter (OM)	
3.5.4 Total Organic Compound (TOC)	27
3.5.5 Particle Size Analysis (PSA)	
3.5.6 Total Nitrogen (TN)	
3.5.7 Total Phosphorus (TP)	
3.6 Statistical Analysis	

3.7 Classification	
4.0 Results	
4.1 In-situ measurements	
4.1.1 Temperature	
4.1.2 pH	35
4.1.3 Dissolved Oxygen	35
4.1.4 Electrical Conductivity	
4.1.5 Turbidity	
4.1.6 Depth and Transparency	
4.2 <i>Ex-situ</i> measurements	40
4.2.1 Five-day Biochemical Oxygen Demand	40
4.2.2 Total Suspended Solids	41
4.2.3 Nitrite-Nitrogen and Nitrate-Nitrogen	42
4.2.4 Total Ammonia Nitrogen	44
4.2.5 Total Nitrogen	45
4.2.6 Soluble Reactive Phosphorus	46
4.2.7 Total Phosphorus	47
4.2.8 Total Sulfide	48
4.2.9 Chemical Oxygen Demand	49
4.2.10 Chlorophyll-a	50
4.3 Sediment Quality	51
4.3.1 pH	51
4.3.2 Water Content	
4.3.3 Organic Matter	
4.3.4 Total Organic Carbon	53
4.3.5 Particle Size Analysis	53
4.3.6 Total Nitrogen	55
4.3.7 Total Phosphorus	56
5.0 Discussion	57
6.0 Conclusion	67
7.0 References	68

Appendices77

List of Abbreviations

%	Percent
°C	Degrees Celcius
AN	Ammoniacal Nitrogen
BOD ₅	Five-day Biochemical Oxygen Demand
Chl-a	Chlorophyll-a
COD	Chemical Oxygen Demand
DO	Dissolved Oxygen
EC	Electrical Conductivity
g	gram
H_2O_2	Hydrogen Peroxide
H_2SO_4	Sulfuric acid
km ²	kilometer square
L	Litre
m	meter
m m ³	meter meter cube
m ³	meter cube
m ³ μg	meter cube microgram
m ³ μg μm	meter cube microgram micrometer
m ³ μg μm μS/cm	meter cube microgram micrometer micro Siemens per centimeter
m ³ μg μm μS/cm mg	meter cube microgram micrometer micro Siemens per centimeter milligram

mL/L	milliLiter per Litre
mm	millimeter
nm	nanometer
NO ₂ -N	Nitrite-Nitrogen
NO ₃ -N	Nitrate-Nitrogen
NTU	Nephelometric Turbidity Unit
NWQS	National Water Quality Standards of Malaysia
рН	Negative Logarithm of Hydrogen Ion Activity
PO ₄ -P	Soluble Reactive Phosphorus
PSA	Particle Size Analysis
rpm	round per minute
SRP	Soluble Reactive Phosphorus
TAN	Total Ammonia Nitrogen
TN	Total Nitrogen
TOC	Total Organic Carbon
TP	Total Phosphorus
TS ²⁻	Total Sulfide
TSI	Trophic State Index
TSS	Total Suspended Solids
Vol	Volume
WQI	Water Quality Index

List of Tables

Table 1:	Description of the sampling stations and GPS coordinates.	16
Table 2:	Date, time, weather condition and observation of the stations.	17
Table 3:	Significant difference of water parameters between July 2014 and February 2015.	34
Table 4:	Mean depth and mean transparency in July 2014 and February 2015.	40
Table 5:	Mean nitrite-nitrogen in all stations in July 2014 and February 2015.	43
Table 6:	Mean nitrate-nitrogen in all stations in July 2014 and February 2015.	43
Table 7:	Comparisons of mean temperature profiling and dissolved oxygen profiling within and among stations in July 2014 and February 2015.	77
Table 8:	Comparisons of mean pH within and among stations in July 2014 and February 2015.	79
Table 9:	Comparisons of mean electrical conductivity within and among stations in July 2014 and February 2015.	79
Table 10:	Comparisons of mean turbidity within and among stations in July 2014 and February 2015.	79
Table 11:	Comparisons of mean five-day biochemical oxygen demand within and among stations in July 2014 and February 2015.	79
Table 12:	Comparisons of mean total suspended solids within and among stations in July 2014 and February 2015.	80
Table 13:	Comparisons of mean total ammonia nitrogen within and among stations in July 2014 and February 2015.	80
Table 14:	Comparisons of mean total nitrogen within and among stations in July 2014 and February 2015.	80
Table 15:	Comparisons of mean soluble reactive phosphorus within and among stations in July 2014 and February 2015.	80
Table 16:	Comparisons of mean total phosphorus within and among stations in July 2014 and February 2015.	81
Table 17:	Comparisons of mean total sulphide within and among stations in July 2014 and February 2015.	81
Table 18:	Comparisons of mean chemical oxygen demand within and among stations in July 2014 and February 2015.	81

Table 19:	Comparisons of mean chlorophyll- a within and among stations in July 2014 and February 2015.	82
Table 20:	pH, water content, organic matter, total organic carbon, total nitrogen and total phosphorus in sediment in July 2014.	82
Table 21:	Sand, silt and clay fractions of the sediment among stations in July 2014.	82

List of Figures

Figure 1:	Location of sampling stations at the study site.	16
Figure 2:	Temperature profile at the three stations in (a) July 2014 and (b) February 2015.	34
Figure 3:	Comparison of mean pH within and among station in (a) July 2014 and (b) February 2015. Same alphabet at same station is not significantly different at 5 % level of significance whereas same number at same depth is not significantly different at 5 % level of significance.	35
Figure 4:	Dissolved oxygen at the three stations in (a) July 2014 and (b) February 2015.	36
Figure 5:	Comparison of mean electrical conductivity within and among station in (a) July 2014 and (b) February 2015. Same alphabet at same station is not significantly different at 5 % level of significance whereas same number at same depth is not significantly different at 5 % level of significance.	38
Figure 6:	Comparison of mean turbidity within and among station in (a) July 2014 and (b) February 2015. Same alphabet at same station is not significantly different at 5 % level of significance whereas same number at same depth is not significantly different at 5 % level of significance.	39
Figure 7:	Comparison of mean BOD ₅ within and among station in (a) July 2014 and (b) February 2015. Same alphabet at same station is not significantly different at 5 % level of significance whereas same number at same depth is not significantly different at 5 % level of significance.	41
Figure 8:	Comparison of mean total suspended solids within and among station in (a) July 2014 and (b) February 2015. Same alphabet at same station is not significantly different at 5 % level of significance whereas same number at same depth is not significantly different at 5 % level of significance.	42
Figure 9:	Comparison of mean total ammonia nitrogen within and among station in (a) July 2014 and (b) February 2015. Same alphabet at same station is not significantly different at 5 % level of significance whereas same number at same depth is not significantly different at 5 % level of significance.	45
Figure 10:	Comparison of mean total nitrogen within and among station in (a) July 2014 and (b) February 2015. Same alphabet at same station is not significantly different at 5 % level of significance whereas same number at same depth is not significantly different at 5 % level of significance.	46
Figure 11:	Comparison of mean soluble reactive phosphorus within and among station in July (a) July 2014 and (b) February 2015. Same alphabet at same station is not significantly different at 5 % level of significance whereas same number	47

	at same depth is not significantly different at 5 % level of significance.	
Figure 12:	Comparison of mean total phosphorus within and among station in (a) July 2014 and (b) February 2015. Same alphabet at same station is not significantly different at 5 % level of significance whereas same number at same depth is not significantly different at 5 % level of significance.	48
Figure 13:	Comparison of mean total sulfide within and among station in (a) July 2014 and (b) February 2015. Same alphabet at same station is not significantly different at 5 % level of significance whereas same number at same depth is not significantly different at 5 % level of significance.	49
Figure 14:	Comparison of mean chemical oxygen demand within and among station in (a) July 2014 and (b) February 2015. Same alphabet at same station is not significantly different at 5 % level of significance whereas same number at same depth is not significantly different at 5 % level of significance.	50
Figure 15:	Comparison of mean chlorophyll- <i>a</i> (Chl- <i>a</i>) within and among station in (a) July 2014 and (b) February 2015. Same alphabet at same station is not significantly different at 5 % level of significance whereas same number at same depth is not significantly different at 5 % level of significance.	51
Figure 16:	Comparison of mean pH in sediment among station in July 2014. Same alphabet at different station is not significantly different at 5 % level of significance.	51
Figure 17:	Comparison of mean water content in sediment among station in July 2014. Same alphabet at different station is not significantly different at 5 % level of significance.	52
Figure 18:	Comparison of mean organic matter in sediment among station in July 2014. Same alphabet at different station is not significantly different at 5 % level of significance.	53
Figure 19:	Comparison of mean total organic carbon in sediment among station in July 2014. Same alphabet at different station is not significantly different at 5 % level of significance.	53
Figure 20:	Comparison of sand fractions in sediment among station in July 2014. Same alphabet at different station is not significantly different at 5 % level of significance.	54
Figure 21:	Comparison of silt fractions in sediment among station in July 2014. Same alphabet at different station is not significantly different at 5 % level of significance.	54
Figure 22:	Comparison of clay fractions in sediment among station in July 2014. Same alphabet at different station is not significantly different at 5 % level of	55

significance.

Figure 23:	Distribution of sand, silt and clay in sediment among station in July 2014.	55
Figure 24:	Comparison of mean total nitrogen in sediment among station in July 2014. Same alphabet at different station is not significantly different at 5 % level of significance.	56
Figure 25:	Comparison of mean total phosphorus in sediment among station in July 2014. Same alphabet at different station is not significantly different at 5 % level of significance.	56
Figure 26:	Scatterplot of water content versus sand.	65
Figure 27:	Scatterplot of clay versus pH.	66

List of Appendices

Appendix A	Mean comparison for <i>In-situ</i> and <i>Ex-situ</i> parameters within and among stations in June 2014 and February 2015.	77
Appendix B	Calibration Curves.	83
Appendix C	Classification of parameters according National Water Quality Standards for Malaysia and Water Quality Index in (a) July 2014 and (b) February 2015.	89
Appendix D	Trophic State Index in (a) July 2014 and (b) February 2015.	90
Appendix E	Water Classes and Uses.	91
Appendix F	National Water Quality Standards for Malaysia (NWQS) and Water Quality Index.	92
Appendix G	Classification of lakes according to Trophic State Index.	94

Water and Sediment Quality of Murum Reservoir, Inundated Plieran River and Danum River

Toh Yu Mei

Resource Chemistry Faculty of Resource Science and Technology Universiti Malaysia Sarawak

ABSTRACT

Dam construction could lead to environmental problems. Three stations, inundated Pleiran River (S1), inundated Danum River (S2) and Murum reservoir (S3) were selected to determine the water quality at 3 depths and sediment quality. Samples collection were in July 2014 and February 2015. S1 which was near to the submerged vegetation area showed the highest BOD₅ (6.77 mg/L), the highest COD (122.03 mg/L) and the highest chl-*a* (12.13 mg/m³) at subsurface in July 2014. S1 had the highest turbidity (50.70 NTU), the highest TAN (0.791 mg/L), the highest SRP (39.09 µg/L) and the highest TS²⁻ (0.09 mg/L) at 20 m depth in February 2015. Total sulfide in both trips showed similar trend at 3 depths. Both sediment texture in S1 and S3 were sandy clay loam; S2 was clay. TSI showed all stations were eutrophic. S2 and S3 at subsurface in July 2014 and S1 and S3 at subsurface in February 2015 showed WQI of Class II (76.5 – 92.7). Other stations fall in Class III (51.9 – 76.5) and IV (31.0 – 51.9) which could affect the healthy growth of aquatic life. Therefore, it is recommended to pre-clear vegetation before the impoundment of the river.

Key words: Murum Dam, water quality, partially-filled, sediment, WQI

ABSTRAK

Pembinaan empangan akan menyebabkan masalah alam sekitar. Tiga stesen, Sungai Pleiran dibanjiri (S1), Sungai Danum dibanjiri (S2) dan takungan Murum (S3) telah dipilih untuk menyelidik kualiti air dalam 3 kedalaman serta kualiti sedimen. Pengambilan sampel pada Julai 2014 dan Februari 2015. S1 yang berhampiran dengan kawasan pokok tenggelam menunjukkan BOD₅ tertinggi (6.77 mg/L), COD tertinggi (122.03 mg/L) dan chl-a tertinggi (12.13 mg/m³) di permukaan pada Julai 2014. S1 mempunyai kekeruhan tertinggi (50.70 NTU), TAN tertinggi (0.791 mg/L), SRP tertinggi (39.09 μ g/L) dan TS²⁻ tertinggi (0.09 mg/L) dalam kedalaman 20 m pada Februari 2015. Total sulfide dalam kedua-dua perjalanan menunjukkan trend yang serupa dalam 3 kedalaman. Tekstur sedimen di S1 dan S3 adalah berpasir tanah liat gembur; S2 adalah tanah liat. TSI menunjukkan semua stesen termasuk dalam eutrofik. S2 dan S3 di permukaan pada Julai 2014 serta S1 dan S3 di permukaan pada Februari 2015 menunjukkan WQI Kelas II (76.5 – 92.7). Stesen lain termasuk dalam Kelas III (51.9 – 76.5) dan IV (31.0 – 51.9) mempengaruhi pertumbuhan sihat kehidupan akuatik. Oleh itu, disyorkan untuk menebang pokok sebelum menaikkan paras air di sungai.

Kata kunci: Empangan Murum, kualiti air, sebahagiannya berisi, sedimen, WQI

1.0 Introduction

Water and sediment quality play an important role in river and reservoir. Dam and reservoir are built for drinking water, hydroelectric generation, flood control, agriculture water supply and recreation (Braga *et al.*, 1998; Verghese, 2001; Vyas, 2001). Dam construction has caused some common environmental problems which are degradation of water quality, severe bio-contamination with bacteria, sediment deposition in reservoir and loss of aquatic life (Eisakhani & Malakahmad, 2009; Gasim *et al.*, 2006). A study by Hu *et al.* (2009) showed that construction of Three Gorges Dam on Yangtze River in China has affected the continuity of the river system for transportation of sediment to downstream of the river. Dam construction in Tasik Chini, Malaysia, showed the water movement becomes less dynamic and affected the fish activities in that area (Gasim *et al.*, 2006).

Dry and wet season and anthropogenic factors contributed to the water and sediment quality. Raining during wet season has caused soil erosion into river and increased the turbidity and total suspended particles in the river (Gasim *et al.*, 2006). Human activities such as logging, agricultural activities, mining, wastewater discharge from residential area are affected the water and sediment quality (Eisakhani & Malakahmad, 2009; Gasim *et al.*, 2006). Marchand *et al.* (2012) reported the water discharge from residential area into Roodeplaat Dam (RD) in city Pretoria, South Africa has affected the aquatic life and people who rely on the water resource. Runoffs of pesticides, herbicides and fertilizers used in agricultural activities have increased the concentration of total nitrate and total phosphorus in the river and led to eutrophication process (Eisakhani & Malakahmad, 2009). Sediment analysis is important as the contaminants are bound to suspended particles and settled with sediments (Mossop *et al.*, 2013).

The construction of Murum dam started in the year 2008 and was expected to operate in the year 2014 (Sovacool & Bulan, 2011; Wong, 2014). As stated by Sarawak Energy (2013), Murum dam which is located in Belaga, Sarawak has a reservoir height of 141 m, catchment area of 2750 km² and it is located upstream of Bakun dam. The main purpose of Murum dam is to generate 944MW of electricity. All the energy generated from Murum dam and Bakun dam was supplied to phosphate, polycrystalline silicon, manganese and aluminium industries (Wong, 2014). People still carry out fishing and hunting activities within the inundated Plieran River, inundated Danum River and reservoir. In addition, logging and conversion of forest to palm oil plantations around the Murum dam catchment area which covered more than 100,000 hectares of land and forest areas may affect the water and sediment quality of Murum reservoir and river (Chidambar, 2014). It is important to carry out water and sediment analysis in Murum dam where literature on the water and sediment quality of this reservoir and river is limited. Therefore, the objective of this study was to determine the water quality of the Murum reservoir and river at different depths of the partially-filled and fully-filled reservoir as well as sediment quality nearby the sampling stations. The water quality was classified according to Water Quality Index (WQI) and Trophic State Index (TSI).

2.0 Literature review

2.1 Water Quality Parameters

2.1.1 Temperature

Temperature plays an important role in a river. Fluctuation of water temperature will affect the photosynthesis rate of plant, physiological responses, metabolic rate and life cycle of aquatic life (Poole & Berman, 2001; McCaffrey, n.d.). McCaffrey (n.d.) stated the temperature change has affected the aquatic life as different organisms have different optimum temperature for them to survive. Temperature is related to dissolved oxygen and depth. Badran (2001) mentioned the higher the water temperature, the lower the amount of dissolved oxygen. Temperature increases as depth decreases (Understanding Water Temperature, n.d.). Water surface gains heat energy directly from sunlight (Understanding Water Temperature, n.d.). Water temperature is raised due to the suspended solids which tend to absorb heat from sunlight (Kumar et al., 2010). In reservoir, thermal stratification happens due to the change in water density with temperature (Chowdhury et al., 2014). Ling et al. (2013b) reported thermocline occurs between 7-11 m in Batang Ai Reservoir. Thermocline layer in reservoir inhibits the oxygen to mix from epilimnion layer to hypolimnion layer (Baharim et al., 2011). Thermal stratification also reduces the DO level and stimulates algae bloom (Chowdhury *et al.*, 2014).

2.1.2 pH

Kwok (2011) stated that pH is used to measure the acidity and alkalinity of water. Presence of more hydrogen ions than hydroxide ions has caused acidity with pH less than 7 in water whereas alkalinity with pH more than 7 happens when the presence of more hydroxide ion than hydrogen ions. pH of most natural waters is between 6.0 - 8.5 (Chapman, 1996). Blood

pH of fish and other organisms have average value of 7.4. Therefore, fish can adapt itself to the external environment and internal body environment. Any change in pH in the river will cause the death of fish (Kwok, 2011). According to Ling *et al.* (2010), the increase of nutrients have induced the formation of algae bloom and lead to increase the pH value. Photosynthesis of algae produced oxygen and increased number of hydroxide ion (Gerardi, 2015). Low value of pH is due to decomposition of organic matter by bacteria in sediment (Nyanti *et al.*, 2012a).

2.1.3 Dissolved Oxygen (DO)

Dissolved oxygen is the concentration of oxygen gas dissolved in water. According to Lawson (1995), 5 mg/L of DO is needed for the healthy growth of aquatic life. Ohio EPA (2014) stated anoxic condition is fall below 0.5 mg/L. Dissolved oxygen is affected by water turbulence, temperature, suspended solid and organic matter. The DO level increases as the water turbulence increases. More oxygen is introduced into the river (Kwok, 2011). Nyanti et al. (2012c) reported high DO level in Kebhor River of Bakun hydroelectric reservoir is due to the water turbulence. Badran (2001) mentioned the higher the water temperature, the lower the amount of dissolved oxygen. According to Chapman (1996), thermal stratification caused the decrease of oxygen level in hypolimnion layer. Increase in suspended solids cause the decrease of DO level in water. Suspended solids tend to trap heat and make oxygen less soluble in water (Understanding Dissolved Oxygen, n.d.). Yisa and Jimoh (2010) mentioned that low DO value is due to organic matter and nutrient runoff from industries. Enriched nutrient in water caused algae blooms. According to Understanding Dissolved Oxygen (n.d.), algae blooms used up oxygen and affect the aquatic lives which require oxygen to live. Decomposition of algae bloom by bacteria leads to depletion of DO in water. This caused the death of aquatic lives. Besides, Chapman (1996) mentioned that decomposition of submerged vegetation during reservoir impoundment reduced the DO level.

2.1.4 Electrical conductivity (EC)

Electrical conductivity should remain below $1000 - 1500 \mu$ S/cm to maintain the aquatic ecosystem (Kwok, 2011). EC is the ability of water to conduct electrical current. EC is affected by the total dissolved solids which are sodium ion, potassium ion, magnesium ion, chloride ion, sulphate ion and carbonate ion. Increase of total dissolved solids are due to the wastewater discharge, fertilizer runoff and flow of water through rock or soil (Das *et al.*, 2006). Hossain *et al.* (2013) mentioned the high rainfall during wet season increased the total dissolved solid. According to Das *et al.* (2006), EC increases with water temperature. Higher water temperature dissolved more mineral from rocks into water (Das *et al.*, 2006). As stated by Chang and Wen (1997), newly impoundment of water in reservoir shows low EC. This is due to the low mineral loads in soil, rock and submerged vegetation in the flooded area.

2.1.5 Turbidity

Kwok (2011) mentioned turbidity is the measured of cloudiness of water due to the total suspended solids in water. Turbidity is measured in nephelometric turbidity unit (NTU). Chapman (1996) stated the high turbidity represents absorption of light by suspended particles in water. High suspended solids in water are due to soil and silt surface runoff, algae growth, heavy rainfall and pollution with organic matter (Chapman, 1996; Kwok, 2011; Nyanti *et al.*, 2012c). High turbidity in the water inhibits the penetration of sunlight into water. Low rate of photosynthesis by phytoplankton has affected the aquatic life by reduced the dissolved oxygen

and food in the water. Suspended solids can clog the gill function of fish and inhibit the development of fish eggs (Minnesota Pollution Control Agency, 2008).

2.1.6 Transparency

Kentucky River Basin Assessment Report (n.d.) stated that the transparency of water is the amount of penetration of sunlight into the water. Transparency of water decreased due to the presence of suspended particles from heavy rainfall, flooding and nutrient runoff. According to Fuller *et al.* (2011), transparency is reduced by the quantity of algae of phytoplankton in the water, tannic acid and turbidity.

2.1.7 Five-day Biochemical Oxygen Demand (BOD5)

According to Yang *et al.* (2009), biochemical oxygen demand is the amount of oxygen used in the oxidation of organic and inorganic compounds by microorganism. Mean BOD₅ value for unpolluted water is 2 mg/L and 3.0 - 6.0 mg/L for aquatic life of European Union standard (Chapman, 1996). BOD₅ value for unpolluted water is 1 mg/L for National Water Quality Standards for Malaysia (DOE, 2012). From previous studies, higher BOD₅ value is due to the untreated wastewater from residential area (Ling *et al.*, 2009; Ling *et al.*, 2013b). Nutrients runoff stimulates algae bloom in the water body. The death of algae contributes to the organic matter and decomposed by bacteria in the water. As a result, oxygen is consumed by bacteria and BOD level increases. Aquatic life is threatened due to the low DO level in the water (*Biological Oxygen Demand – Overview*, n.d.).

2.1.8 Total Suspended Solids (TSS)

Total suspended solids are the concentration of suspended particles in water (Queenan *et al.*, 1996). Kumar *et al.* (2010) stated the suspended solids tend to absorb heat from sunlight and increased the water temperature. High suspended solids in water are due to soil and silt surface runoff, algae growth, heavy rainfall and pollution with organic matter (Chapman, 1996; Kwok, 2011; Nyanti *et al.*, 2012c). Ling *et al.* (2010) reported high level of total suspended solids at Santubong River due to the new road development and construction in progress. Total suspended solids have affected the DO and BOD₅ level in the water (Nyanti *et al.*, 2012a).

2.1.9 Nitrite-Nitrogen (NO₂-N) and Nitrate-Nitrogen (NO₃-N)

Camargo and Ward (1992) mentioned NO₃-N is the stable form of nitrogen whereas NO₂-N contains nitrogen with unstable oxidation state and NO₂-N is easily being oxidized into NO₃-N. NO₂-N is more toxic than NO₃-N. NO₃-N is least toxic but it occupies high concentration in water. Extensive used of fertilizers in fish farming and effluent from wastewater treatment plants contribute high level of NO₂-N and NO₃-N (Boyd, 2003; Gasim *et al.*, 2006). U.S. Environmental Protection Agency (2007) stated NO₂-N and NO₃-N which are soluble in water and move into river through leaching or runoff. Elevated nutrient level causes eutrophication process in the river and affects the aquatic life. Studies show the nutrients runoff into groundwater have affected drinking water source. Methemoglobinemia or known as blue baby syndrome occurs in infants who drink the contaminated water (U.S. Environmental Protection Agency, 1991).